

7. A student massed a rock and determined it to have a mass of 31.270 g. Then, the student placed some water in a graduated cylinder. The initial liquid level was 25.6 mL. The rock was carefully placed in the graduated cylinder, and the liquid level rose to 39.5 mL. What was the density of the rock?
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8. The liquid level in a graduated cylinder is 51.20 mL. After adding a piece of porcelain, the liquid level of the water is 76.17 mL. The porcelain piece was removed from the graduated cylinder and carefully dried off. The porcelain piece was found to have a mass of 42 g. What is the density of the porcelain piece?
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Group Processing Report

	Yes	Sometimes	No
We accomplished our assignment.			
We stayed with our group.			
We gave praise to all members of the group when needed.			
We helped all group members when needed.			
We shared our materials.			
We feel all group members contributed to this assignment.			

We learned:

We re-learned:

Areas in which we could show improvement:

Comments and suggestions:

Group Signatures:
