

Chemistry CP

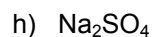
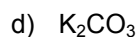
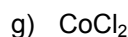
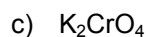
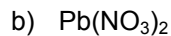
Section:

Lab: Precipitates and Their Formulas

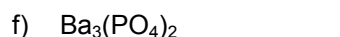
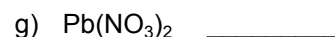
Name:

Prelab Questions

1. List the cation and the anion which make up the following compounds:



2. Using your solubility table, predict whether the following combinations are soluble (S), slightly soluble (SS), or insoluble (I).



3. You combine the following solutions and obtain a precipitate. Write balanced equations for each of the following reactions. Using your solubility table and ion chart, predict the formula of the precipitate (i.e., the product that is NOT soluble in water).

