

6. What is the mass in grams of a .28 mole sample of gold(III) cyanide?
7. Calculate the mass of a sample of iron(II) hydroxide which contains 8.36×10^{23} formula units of iron(II) hydroxide.
8. How many formula units of gold(III) cyanide are present in a 10.76 g sample of gold(III) cyanide?

The purpose of this assignment was to:

Did I:	Circle the appropriate response:		
Explain how I did the problems?	Always	Sometimes	Rarely
Listen while my partner explained?	Always	Sometimes	Rarely
Give my partner positive support?	Always	Sometimes	Rarely
Stay on task during the assignment?	Always	Sometimes	Rarely
Use encouraging and polite words?	Always	Sometimes	Rarely
Record my work on the paper?	Always	Sometimes	Rarely
Demonstrate an understanding of the material?	Yes	No	

Signatures:

Comments: